Alkoxometalates

A possible synthesis route to homonuclear aluminium sesquialkoxides is the reaction of equimolar amounts of aluminium with an alcohol in xylene under reflux conditions [eqn. (1)]. Al($O^{c}Hex$)₃ is trimeric in the solid state (fig. 1)

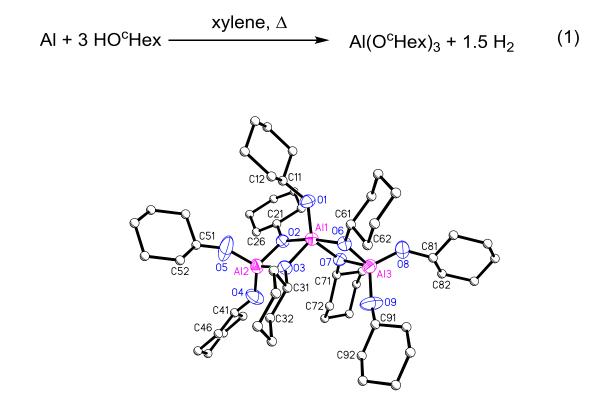


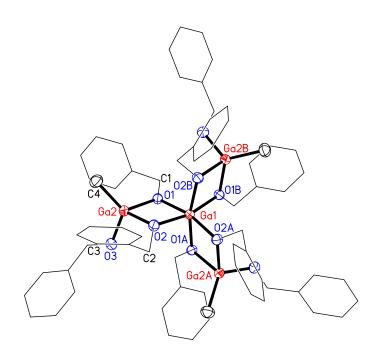
fig. 1

Reaction of Trimethylgallane with PhCH₂OH leads to a sesquialkoxide according eqn. (2).

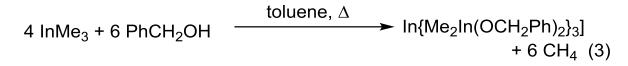
4 GaMe₃ + 9 PhCH₂OH
$$\longrightarrow$$
 Ga{Me₂Ga(OCH₂Ph)₃}₃]
+ 9 CH₄ (2)

The structure of this sesquialkoxide is shown in fig. 2.





With ^cHexOH and InMe₃ another type of sesquialkoxide was formed accoding eqn. (3).



The structure is similar, shown in fig. 3.

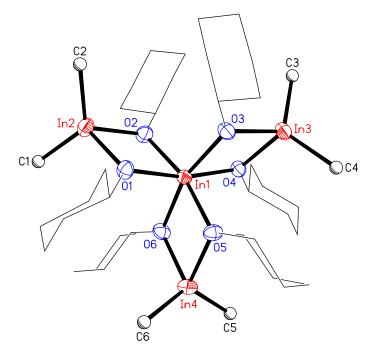


fig. 3

Me₃Ga and two equivalents of alcohol gives MeGa(OR)₂ according eqn. (4) which are polymeric in the solid state (fig. 4).

GaMe₃ + 2 ROH
$$\rightarrow$$
 1/n [{MeGa(OR)₂}₂]_n + 2 CH₄ (4)
R = Et, ⁿBu

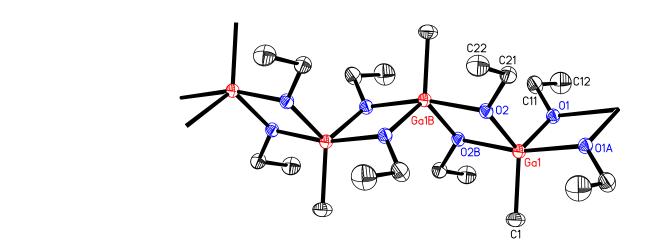


fig 4

InMe₃ undergoes in PhCH₂OH under reflux conditions the reaction to a complex with five In atoms (eqn. 5).

InMe₃
$$\xrightarrow{\text{PhCH}_2\text{OH}, \Delta}$$
 [(Meln)₅(OCH₂Ph)₈(O)] (5)
- CH₄

The complex is shown in fig. 5.

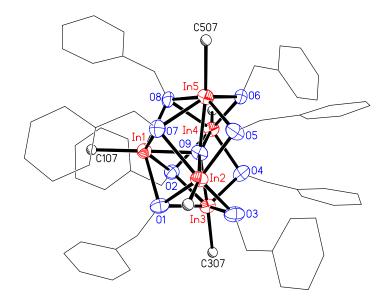


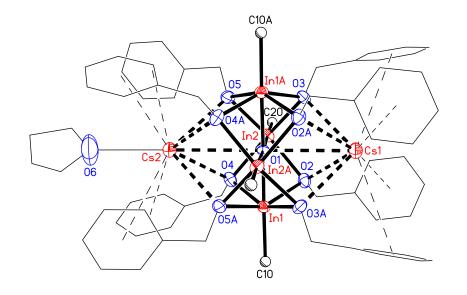
fig. 5

Reaction of [(MeIn)₅(OCH₂Ph)₈(O)] with cesium in organic solvents under ultrasonic conditions leads to a Cs complex according eqn. (6).

$$(Meln)_5(OCH_2Ph)_8(O)] + 2 Cs$$
 1. toluene, 20 °C, ultrasound
2. THF

 $[Cs{Cs(THF)}{Meln(OCH_2Ph)_2]_4(O)}] (6)$

In the cesium comlex the Cs+-Ions are coordinated by oxygen atoms and the π -systems of the OCH₂Ph ligands (fig. 6).



Literature:

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